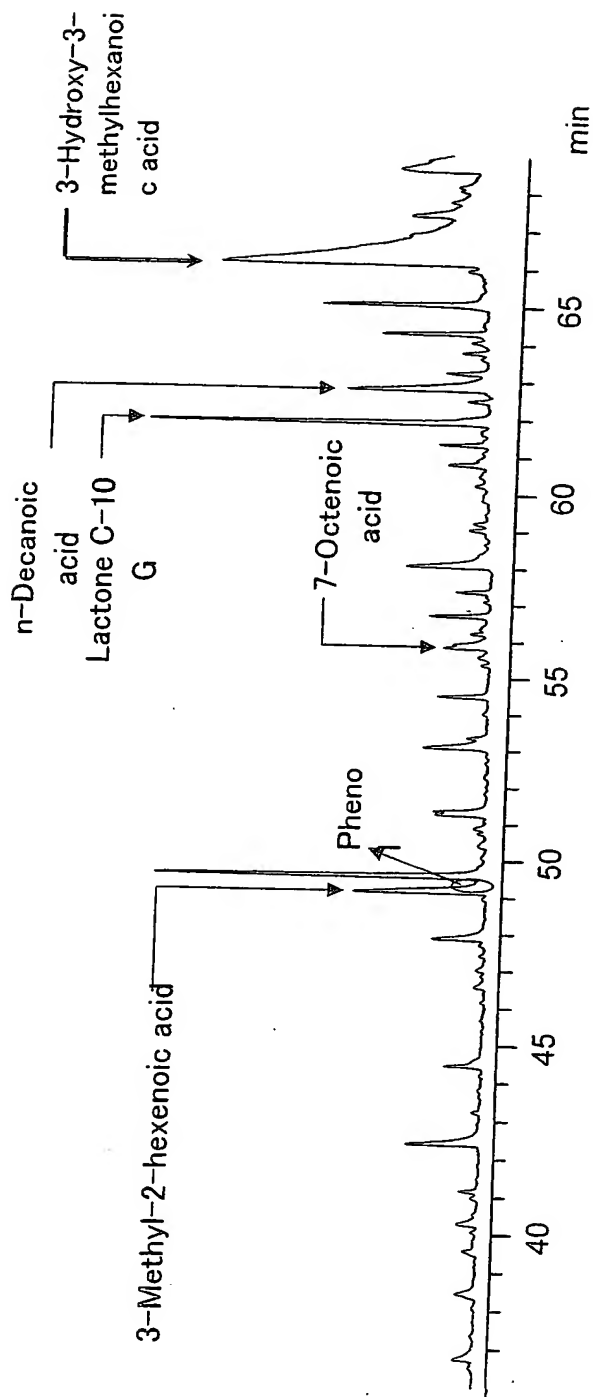


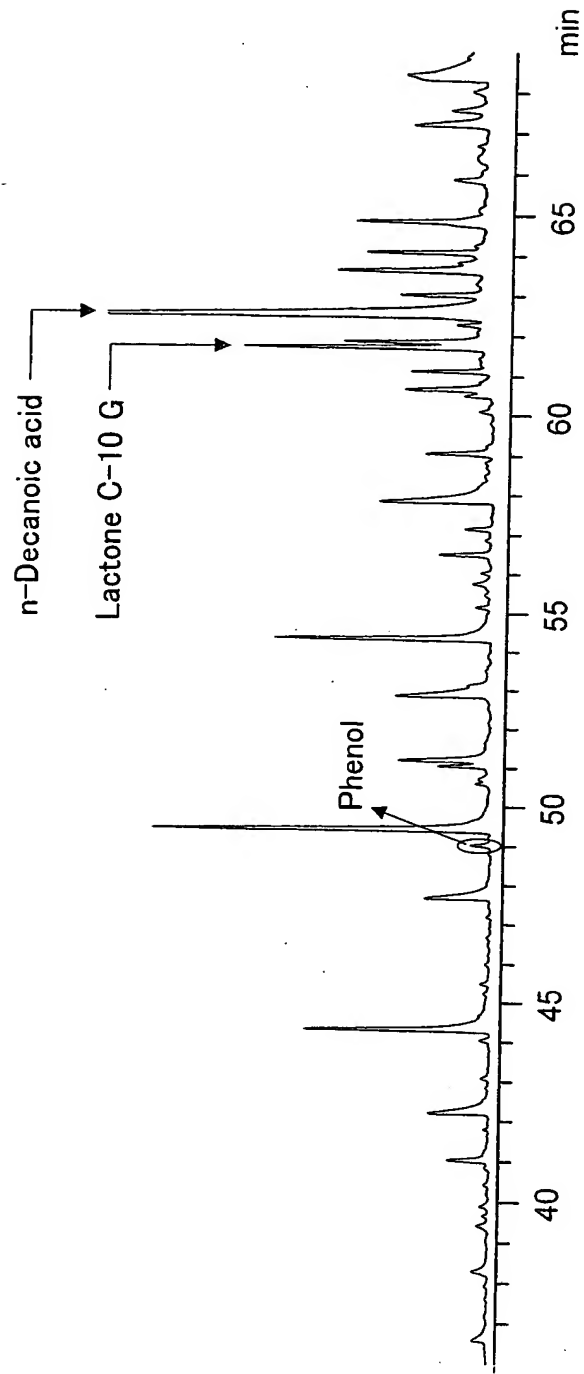
FIG. 1



Total ion chromatogram of an acid extract of perspiration
of a person having an apocrine odor

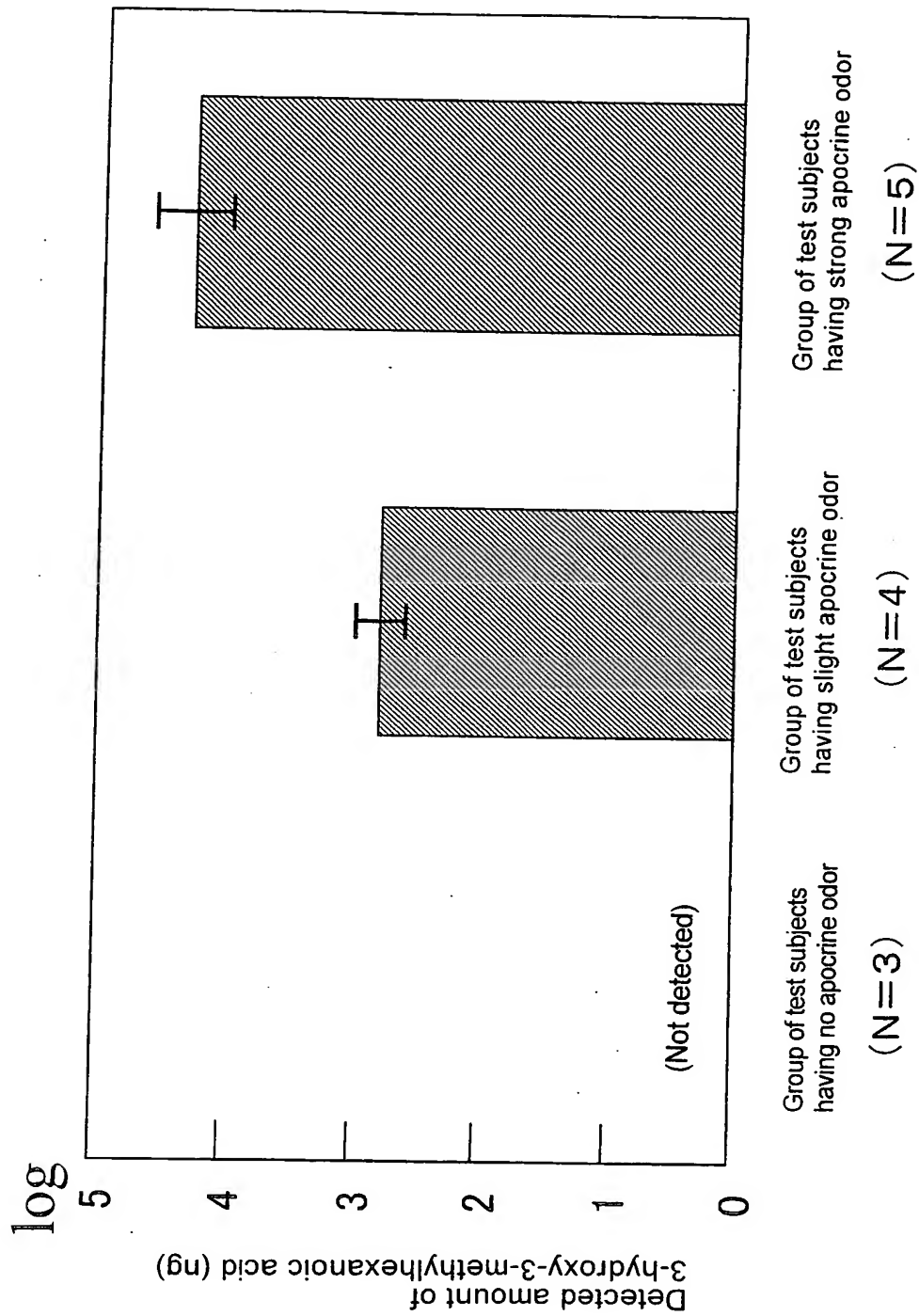


FIG. 2



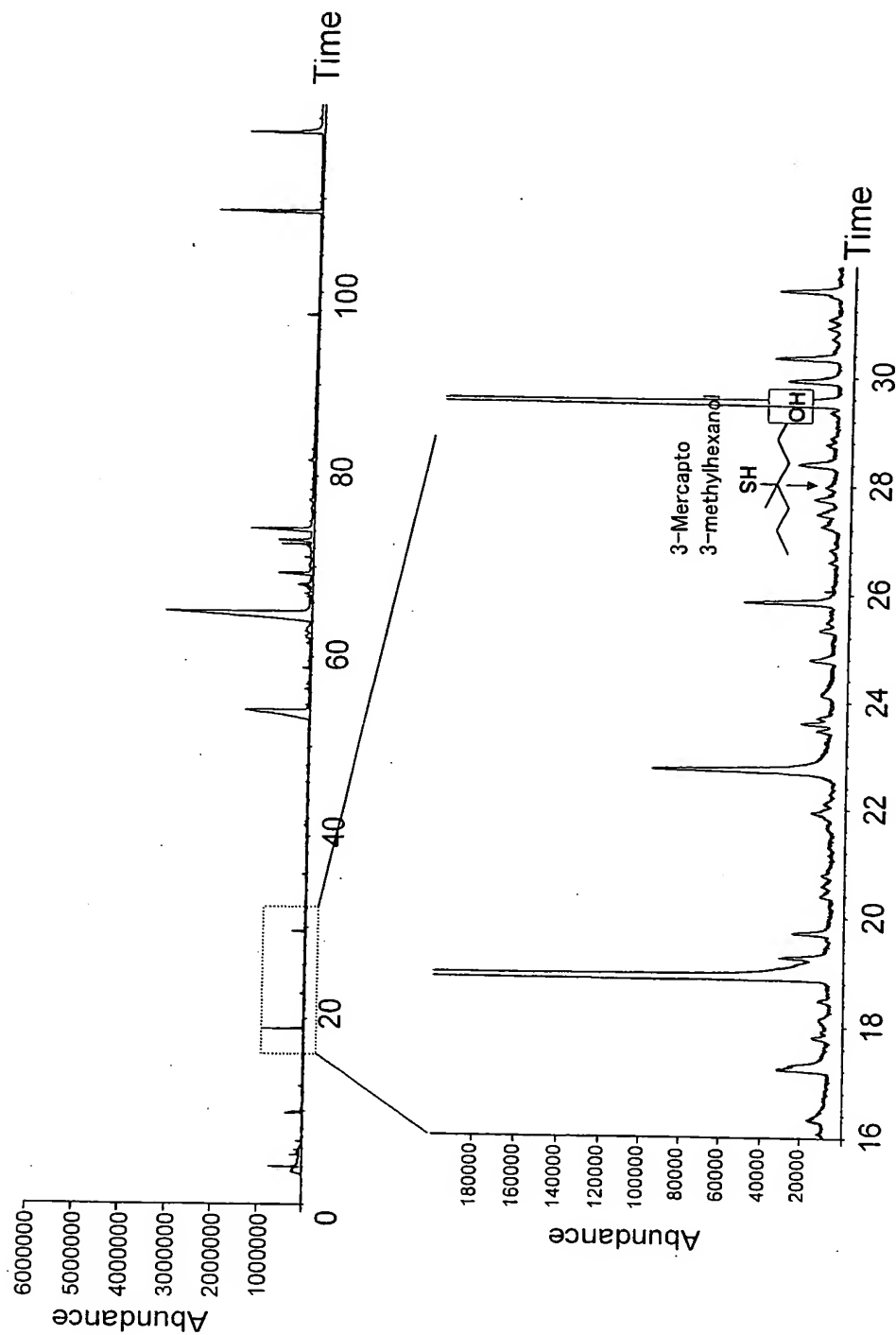
Total ion chromatogram of an acid extract of perspiration
of a person having no apocrine odor

FIG. 3



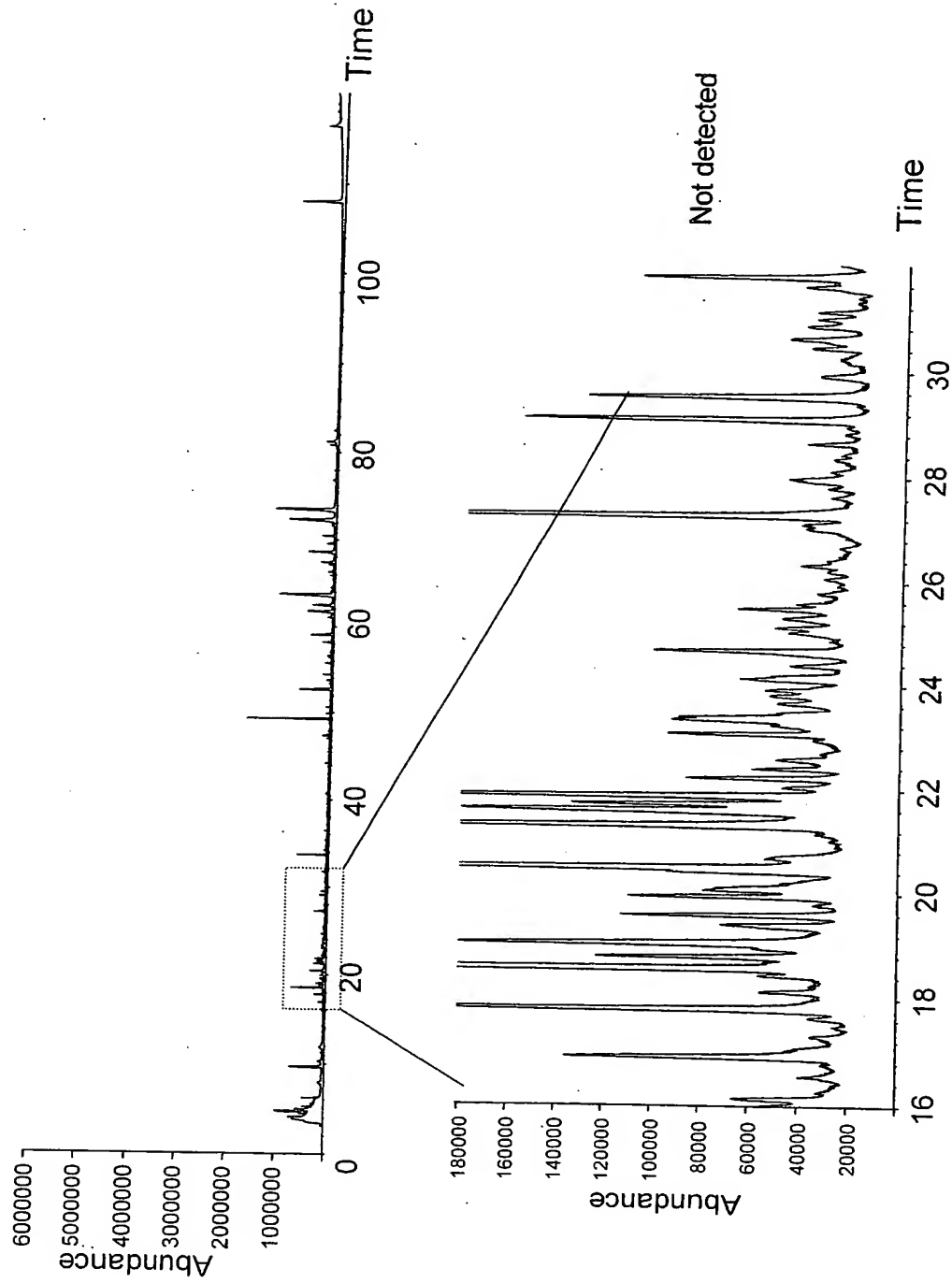
Relationship between 3-hydroxy-3-methylhexanoic acid and an apocrine odor

FIG. 4



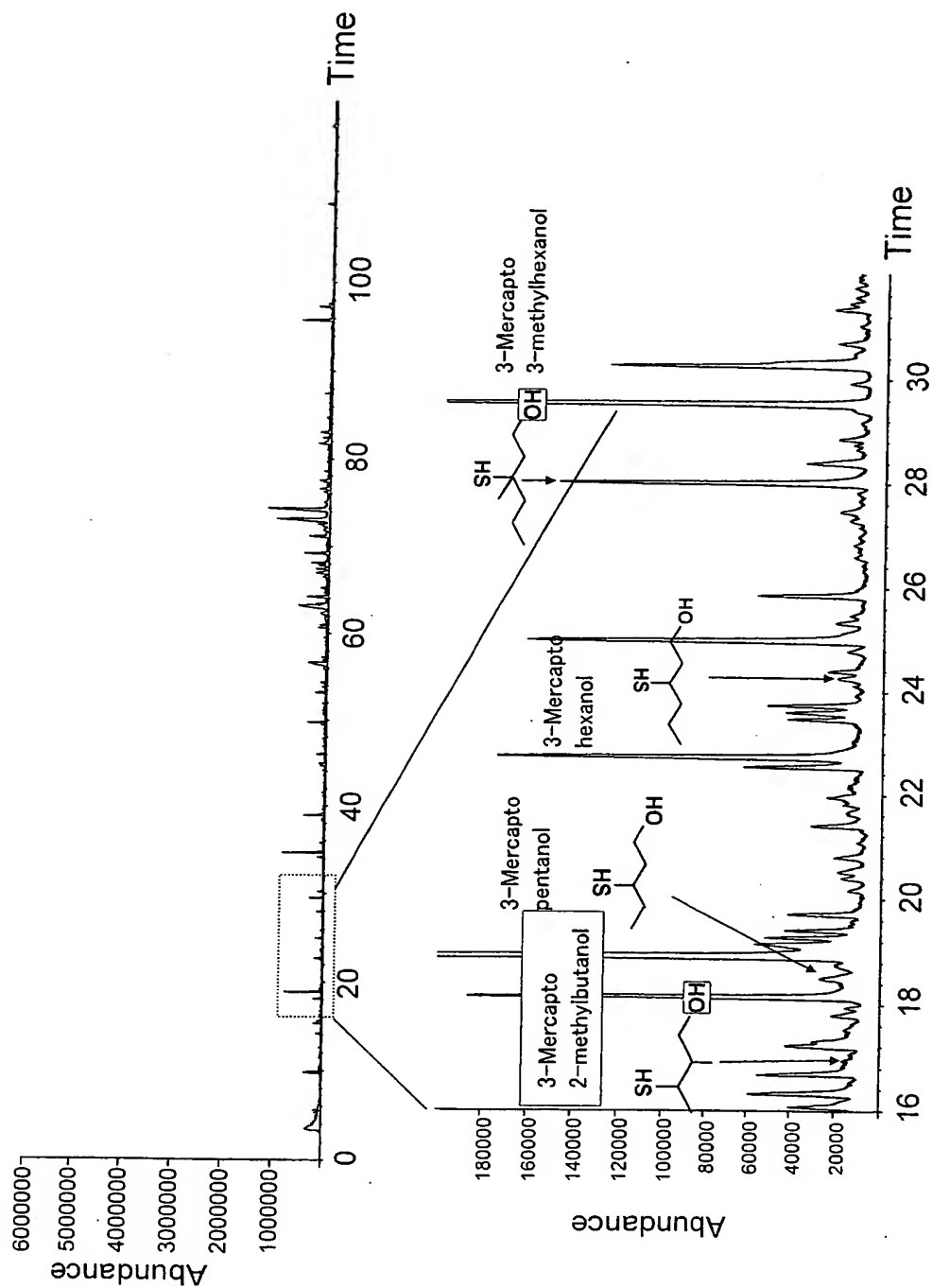
Total ion chromatogram of volatile components in perspiration
of a person having an apocrine odor

FIG. 5



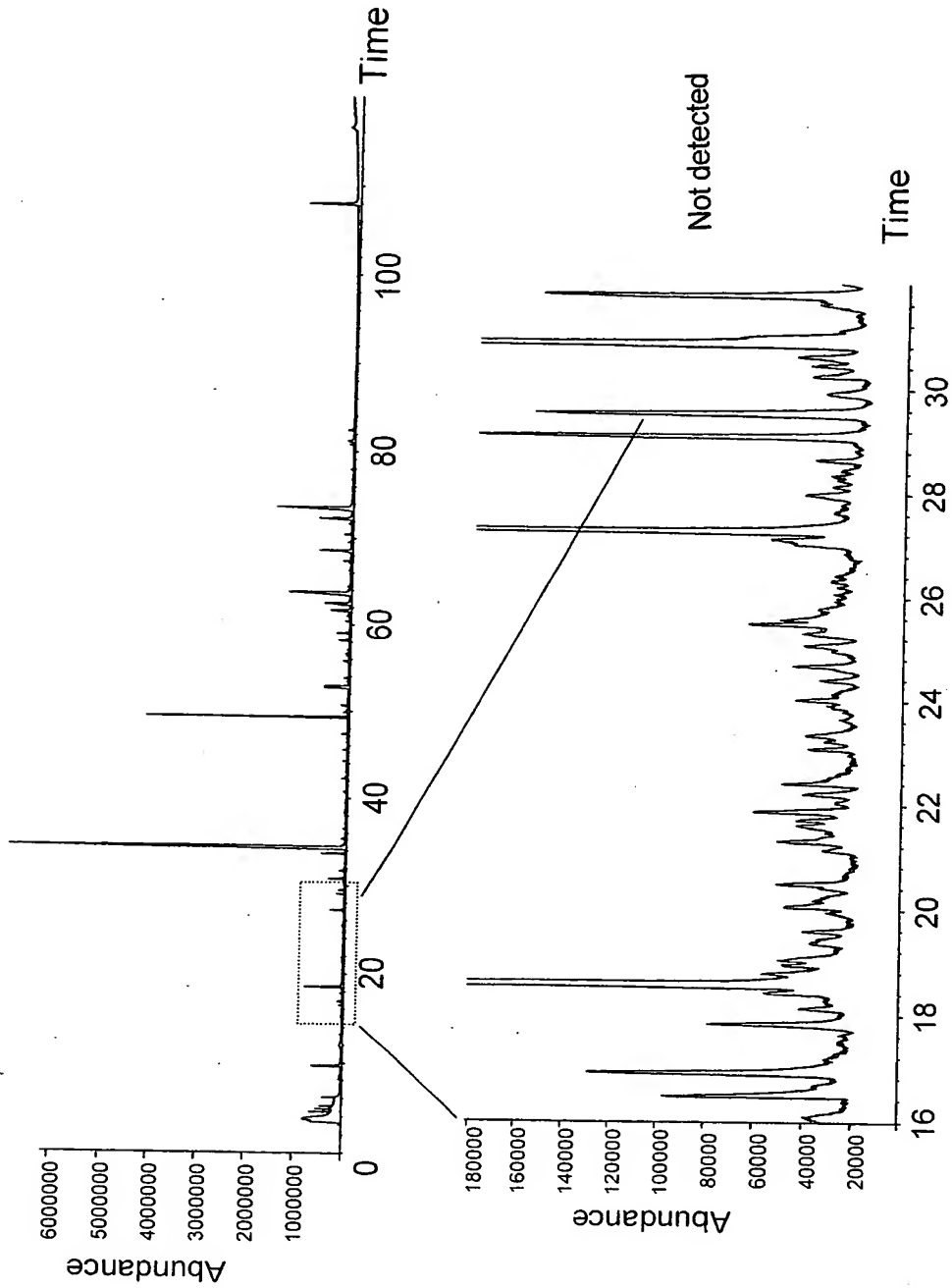
Total ion chromatogram of volatile components in perspiration
of a person having no apocrine odor

FIG. 6



Total ion chromatogram of volatile components in incubated perspiration
of a person having an apocrine odor

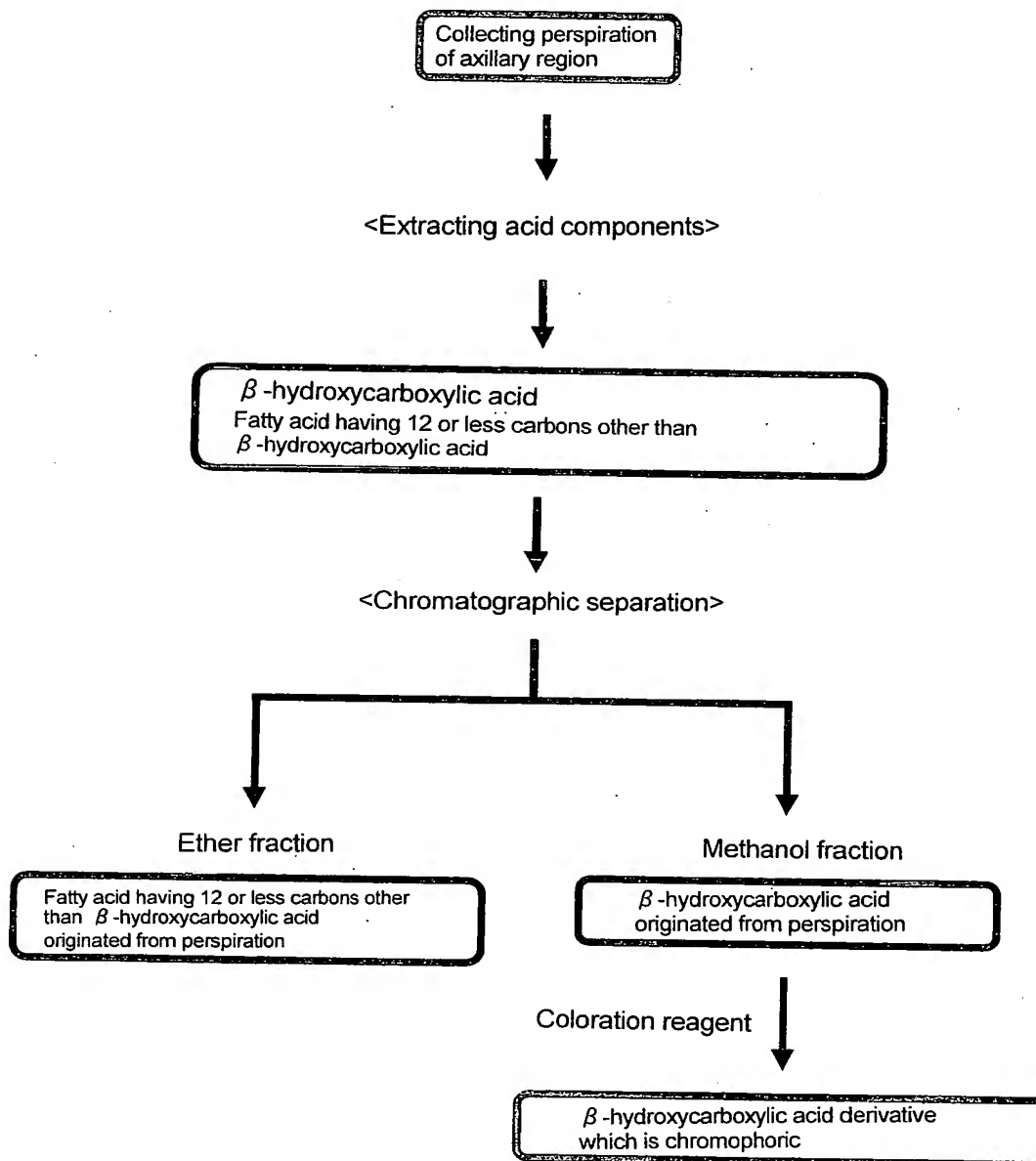
FIG. 7



Total ion chromatogram of volatile components in incubated perspiration
of a person having no apocrine odor

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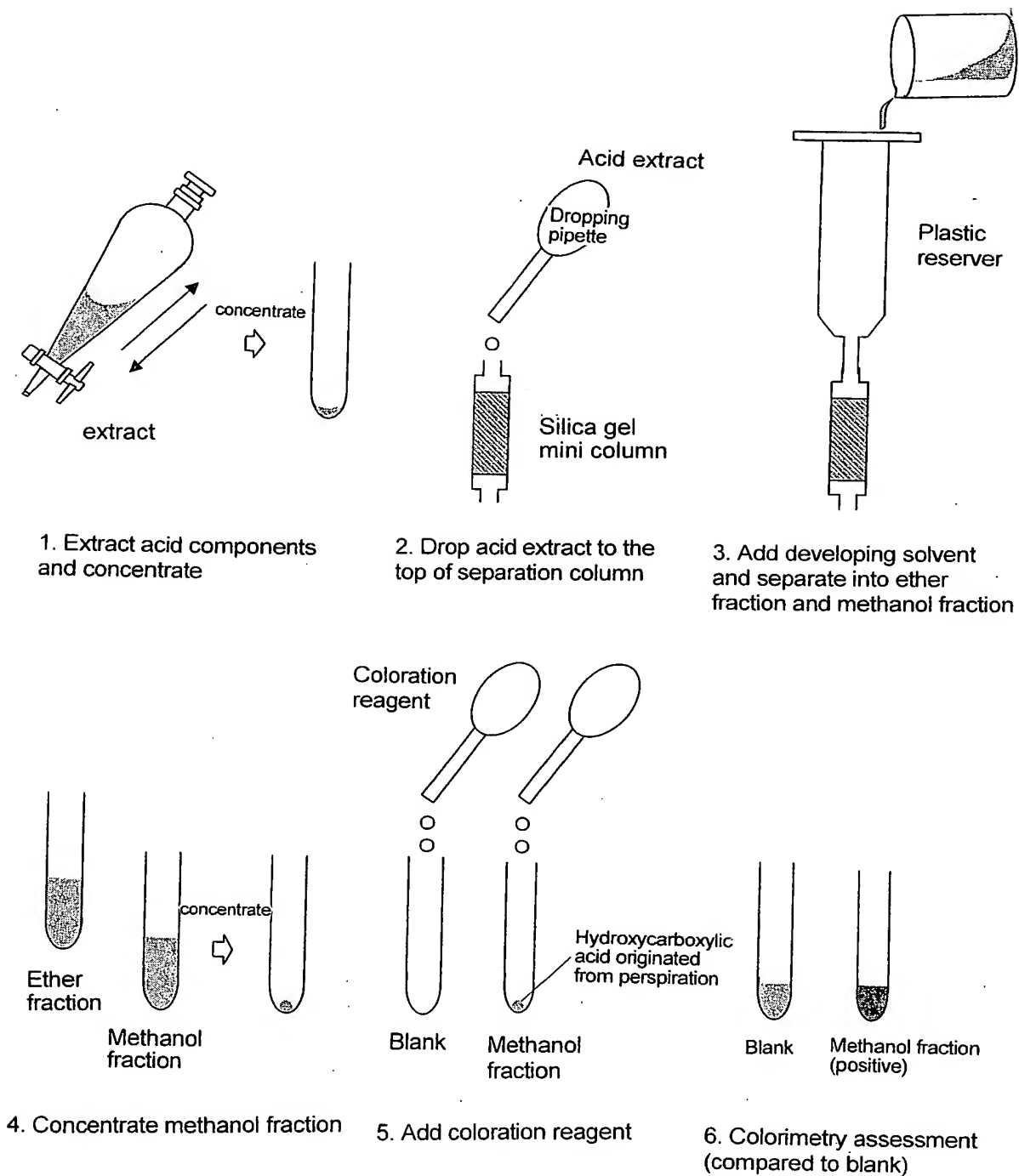
FIG. 8



Quick and convenient method for examining body odor
utilizing the coloration reaction

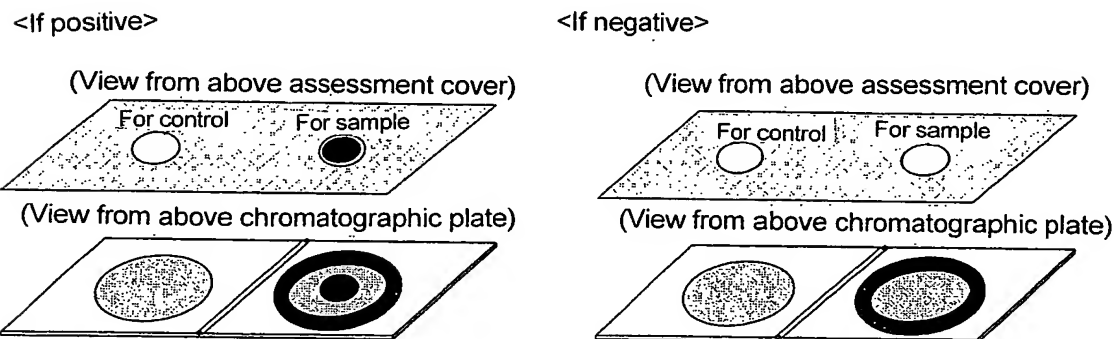
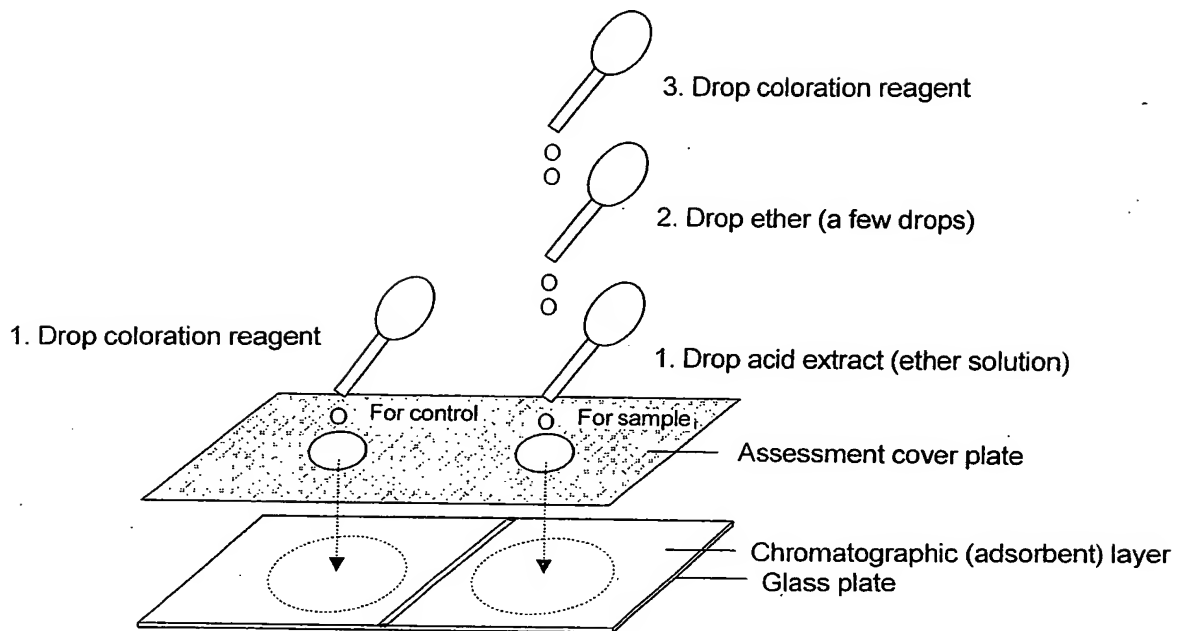
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FIG. 9



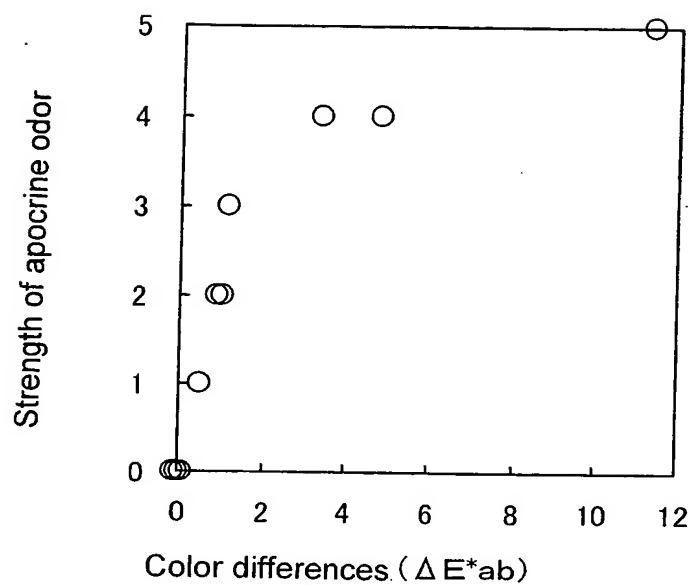
Example (1) of how kit for conveniently accessing body odor is used

FIG. 10



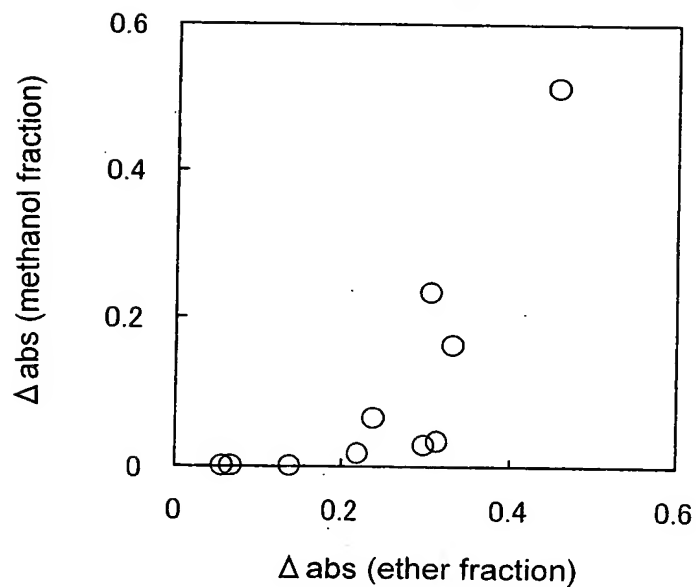
Example (2) of how kit for conveniently accessing body odor is used

FIG. 11



Relationship between strength of
apocrine odor and color differences

FIG. 12



Absorbance of methanol
fraction and ether fraction